

## Standardized Test Prep Chemistry Chapter 11

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**Standardized Test Practice**

World of Chemistry 28 Copyright © Houghton Mifflin Company. All rights reserved. Name \_\_\_\_ Section \_\_\_\_ Date \_\_\_\_ 10. A student drew the following electron box ...

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**Standardized Test Practice-English**

Answers by Chapter. Chapter 1 - Introduction to Chemistry. Chapter 2 - Matter and Change. Chapter 3 - Scientific Measurement. Chapter 4 - Atomic Structure. Chapter 5 - Electrons in Atoms. Chapter 6 - The Periodic Table. Chapter 7 - Ionic and Metallic Bonding. Chapter 8 - Covalent Bonding.

**Chapter 4 - Atomic Structure - Standardized Test Prep...**

Answers by Chapter. Chapter 1 - Introduction to Chemistry. Chapter 2 - Matter and Change. Chapter 3 - Scientific Measurement. Chapter 4 - Atomic Structure. Chapter 5 - Electrons in Atoms. Chapter 6 - The Periodic Table. Chapter 7 - Ionic and Metallic Bonding. Chapter 8 - Covalent Bonding.

**Chapter 8 - Covalent Bonding - Standardized Test Prep...**

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Chapter 6 Standardized Test Practice 1. The actual number of atoms of each element contained in a molecule is given by the molecule's: A. empirical formula. C. mass percent formula. B. molecular formula. D. percent composition. 2. What is the molar mass of the sodium sulfate Na2SO4? F. 71.06 g/mol G. 94.05 g/mol H. 142.05 g/mol J. 215.13 g/mol 3.

**Chapter 6 Standardized Test Practice - web1.tvusd.k12.ca.us**

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