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**Molarity,
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Density \u0026
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**Volume Percent -
Chemistry** *What's
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Introduction

Percentage

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Calculations

What are

Solutions?

Molality and

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molality |

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Many companies, including pharmaceutical companies, keep track of the concentration of a solution by measuring its molarity – a ratio of the

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Answers
Number of solute particles to the volume of the solution. In this activity you will learn about molarity and how to represent concentration quantitatively.

Model 1 –
Lemonade
Mixtures*

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Molarity POGIL Key | Molar Concentration | Solution

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Answers
is molarity (M).

Molarity is defined as the number of moles of solute (mol) divided by the total volume (V) of the solution in liters (L).

Molarity = moles of solute per liter of solution ($M = \text{mol} / \text{L}$).

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Answers -

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of the new
solution?

(Answer: 9-801

HN03) = ML (

0.8M -N 3-30.00

ml, of water is

added to 20.00

ml, of 0.0500 M

sodium

carbonate. What

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is the molarity
of sodium
carbonate in the
new solution?

(Answer: 0.0200
M) (20.00 m C
(350. 00 casF)

4. What volume
of 0.100 M NaCl
is needed to
make 5.00 ml, of
0.200 M
solution?

(Answer: 0.01

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(0.100 M) V)

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