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gases Identification of cations
(positive ions) Identifying metal
cations by using flame tests The

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principle here is that the salts of some metals will impart a colour to a non-luminous Bunsen flame. Test 1. The technique is first of all to clean the end of a piece of platinum or nichrome wire by dipping it into

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$\text{N}_2\text{O}_4(\text{g}) \rightleftharpoons 2 \text{NO}_2(\text{g})$ If a reaction vessel initially contains an N_2O_4 concentration of 0.0500 M at

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