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reaction caused by absorption of ultraviolet (wavelength from 100 to 400 nm), visible light (400–750 nm) or infrared radiation (750–2500 nm).

[Photochemistry - Wikipedia](#)

The fascinating field of inorganic photochemistry is extremely diverse. This chapter discusses some general principles governing light-induced properties of metal-containing molecular compounds.

[Inorganic Photochemistry | SpringerLink](#)

Research in inorganic photochemistry has expanded enormously in scope and importance, especially in the past decade. The development was encouraged by Balzani and Carassiti in their exhaustive monograph (I) on the photo-chemistry of coordination compounds that just preceded the decade.

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Some of the early quantitative history of inorganic photochemistry is recalled and some of the early experiences of the writer. Conceptual landmarks in the progress to the present are outlined. The role of the Italian school of photochemistry in this development is noted.

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The study of the light-induced behavior of various metal compounds. The physical and chemical properties of substances are generally altered by the absorption of light. Typical metal compounds have a characteristic number (coordination number) of molecules or ions (ligands) directly bonded to the metal center.

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This monograph/reference focuses on those subjects that are considered essential to an understanding of inorganic photochemistry. Graduate students with a background in physical chemistry will find that the quantum mechanical treatments related to the principles of spectroscopy and chemical dynamics are readily accessible.

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