

Chapter 1 Matter And Change

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CHAPTER 1 REVIEW Matter and Change SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a Technological development of a chemical product often (a) lags behind basic research on the same substance. (b) does not involve chance discoveries. (c) is driven by curiosity. (d) is done for the sake of learning something new.

1 Matter and Change - HUBBARD'S CHEMISTRY

8 CHAPTER 1 Before Chemistry Until the chemical revolution of the seventeenth and eighteenth cen-turies, most theories about matter were based on the ideas of the ancient Greek philosopher Aristotle. He postu-lated that all matter consisted of four elements: earth, water, air, and fire. In turn, each of these elements exhibited two of four fundamental

CHAPTER 1 Matter and Change

Chapter 1: Matter and Change Section 1-1: Chemistry Is a Physical Science _____ Pacing Regular Schedule: with lab(s): 2 days without lab(s): 1 day Block Schedule: with lab(s): 1 day without lab(s): 1/2 day Objectives 1. Define chemistry. 2. List examples of the branches of chemistry. 3.

Chapter 1: Matter and Change

a characteristic that can be observed or measured without changing the identity of a substance. physical change. a change in a substance that does not involve a change in the identity of a substance. state of matter. is a physical change of a substance from one state to another. 3 states of matter are solid, liquid, gas.

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All matter is made up of particles. Particles of matter are always in motion. The kinetic energy (speed) of these particles increases as temperature increases. The temperature and kinetic energy of the particles determines the state of matter

Unit 1: Chapter 1 [Matter and Change]

Chapter 1 Section 2 MATTER and Its Properties Matter [Anything that has MASS and VOLUME. ATOMS and MOLECULES the basic building blocks of matter. ATOM-- the smallest unit of an element that still maintains the properties that element. ELEMENT[a pure substance made up of only one type of atom.

MATTER and CHANGE Chapter 1 Section 1

the study of the properties and changes of matter and their relation to energy (properties and charges of matter via energy) ex. understanding at the molecular and atomic level of how materials behave and chemical reactions occur)

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A physical science which studies the composition of matter and the changes which matter can undergo. (ex: sugar- composition= O, C, H []> chemical- reacting with acid, physical- crushing & dissolving. Matter. Anything which has mass and has volume. (Takes up space). (ex: desk, binder, bookbag, pencil, humans.)

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Two factors responsible for the change of state of matter are: change in □ Temperature □ Pressure Question 4. State the main postulates of kinetic theory of matter. Solution: The main postulates of the theory are: 1. Matter is made up of very small particles. 2. The constituent particles of a kind of matter are identical in all respects. 3.

Chapter 1 - Matter

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Chapter 1: Matter and Change (Chem in 15 minutes or less ...

Chapter 1: Matter and Change. Lesson 1-1: Chemistry is a Physical Science. Objectives: 1.1 Define chemistry. 1.2 List examples of the branches of chemistry. 1.3 Compare and contrast basic research,...

ATA Chemistry - 1 Matter and Change

Chemistry Chapter 1 Matter and Change Chemistry is the study of matter-1. its composition 2. its structure 3. its properties 4. the changes that it undergoes Study of different branches of chemistry 1. organic- contains carbon 2. inorganic- not organic- usually no carbon 3. physical-properties & changes in matter & relation to energy 4. analytical- composition 5. biochemistry- substances and processes in living things 6. theoretical- design & predict properties of new compounds- uses math ...

Chapter 1 Study Guide.doc - 1 Chemistry Chapter 1 Matter ...

Name _____ Period _____ Due Date _____ 1 Chemistry Chapter 1: Matter and Change Study Guide Section 1: Chemistry is a Physical Science

Chemistry Chapter 1: Matter and Change Study Guide

CHAPTER 1 Matter and Change Chapter 1: Matter and Change Lesson 1-1: Chemistry is a Physical Science Objectives: 1.1 Define chemistry. 1.2 List examples of the branches of chemistry. 1.3 Compare and contrast basic research, applied research, and technological development. Vocabulary: chemistry chemical decant filter

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