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Q. Deep-sea divers use a breathing mixture of helium, oxygen, and carbon dioxide. What is the partial pressure of oxygen if the total pressure of the mixture is 760 mm Hg, the pressure exerted by the helium is 609 mm Hg, and the pressure exerted by the carbon dioxide is 1 mm Hg?

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Think about these steps to answer the question: Identify the system - P.E.). goes from liquid PE change, In this This is an type of change, the system (the water) heat from the surroundings. Identify the surroundings - This reaction is When Act E is high, the reaction is (slow, fast). Sketch a diagram of these reactions: thermic. faster, exothermic

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